

1,4-SUBSTITUTED INDOLES: A POTENT AND SELECTIVE CLASS OF ANGIOTENSIN II RECEPTOR ANTAGONISTS

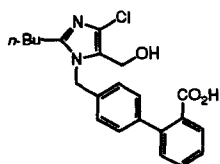
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Abstract: The syntheses and pharmacological activity of a series of 1,4-substituted indoles which function as nonpeptidic antagonists of the angiotensin II (AII) receptor are described. Compounds in this series are orally active and demonstrate long lasting antihypertensive activity.

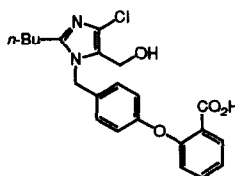
Angiotensin converting enzyme (ACE) inhibitors, such as captopril and enalapril, have been demonstrated to be clinically effective in the treatment of hypertension and congestive heart failure.¹ These compounds regulate the renin angiotensin system (RAS) by blockade of the biosynthesis of the potent endogenous vasoconstrictor angiotensin II (AII). An alternative strategy to regulate the RAS involves blockade of the action of AII at its receptor by an antagonist. Recently, the development of AII receptor antagonists has been an area of active interest and several clinical candidates have emerged.²

We describe here our work to develop a series of 1,4-substituted indoles which are nonpeptidic antagonists of the AII receptor. Our initial strategy to develop a novel, potent, and selective AII receptor antagonist was influenced by the data reported by workers at DuPont^{2a,2e,3} for compounds 1-3. Compounds 1-3 have nearly equal affinity for the AII receptor, but structurally have different linkers between the two 6-membered aryl rings which vary in length from one bond (1) to three bonds (3). This data suggested to us that compounds containing other linkers, particularly derivatives with more conformational constraint, might also be effective antagonists of the AII receptor.



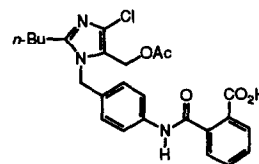
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IC₅₀ (μM) = 0.23



2

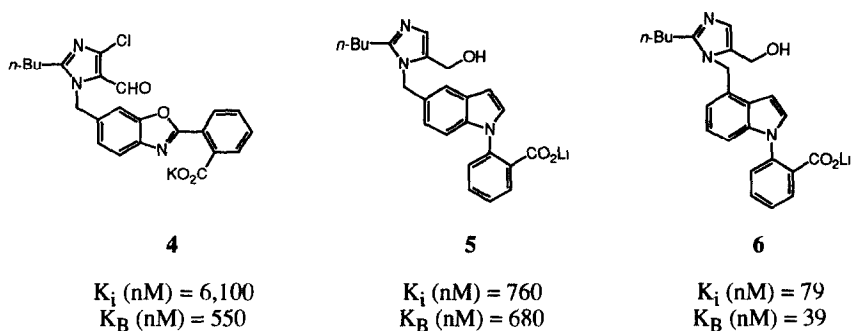
IC₅₀ (μM) = 0.40



3

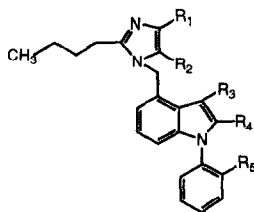
IC₅₀ (μM) = 0.28

We elected to examine the [4.3.0] aromatic ring system⁴ as a replacement for both the central aryl ring and the linker in compounds 2-3. To study different substitution patterns, compounds 4-6 were prepared. From this effort, the 1,4-indole substitution pattern present in 6 emerged as superior.^{5,6}



Having identified a prototype, our efforts focused next on optimization of the 1,4-substituted indole template (**Table 1**). Replacement of the carboxylic acid in **6** with a C-linked tetrazole and incorporation of chlorine at the 4-position of the imidazole⁷ provided compound **7**, a derivative which was significantly more potent than our initial lead **6**. A further enhancement in potency was achieved by oxidation of the hydroxymethyl residue to a carboxylic acid (**8**).⁷ Additional modification of the imidazole ring demonstrated the need for substitution of the 4-position with an electron withdrawing group. Hydrogen at the 4-position caused a dramatic decrease in potency possibly due to the attendant increased basicity of the imidazole ring (**9** vs. **8**), whereas a trifluoromethyl analog was moderately potent (**10** vs. **8**).

Table 1.

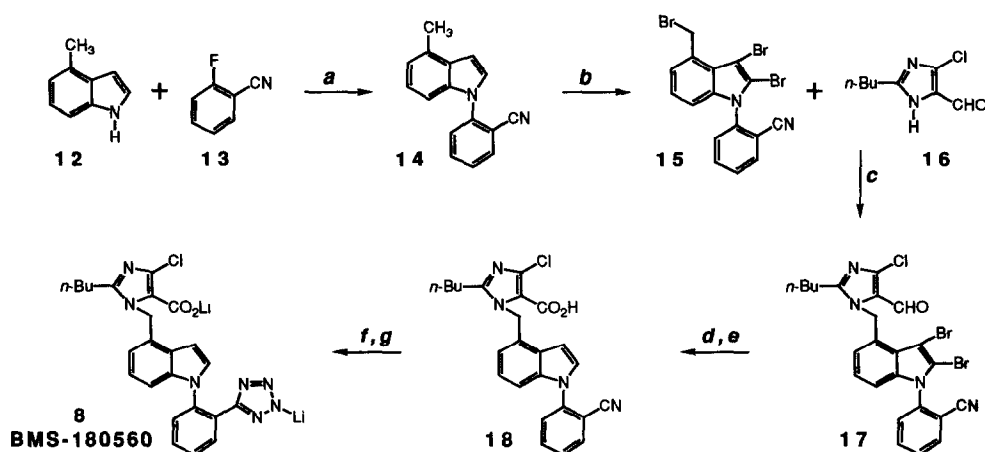


Compound	R ₁	R ₂	R ₃	R ₄	R ₅	K _i (nM)	K _B (nM)
6	H	CH ₂ OH	H	H	CO ₂ Li	79	39
7	Cl	CH ₂ OH	H	H	CN ₄ Li	6.3	0.90
8 (BMS-180560)	Cl	CO ₂ Li	H	H	CN ₄ Li	0.8	0.068
9	H	CO ₂ H	H	H	CN ₄ H	3600	270
10	CF ₃	CO ₂ Na	H	H	CN ₄ Na	79	0.12
11	Cl	CO ₂ Li	Br	Br	CN ₄ Li	840	300

The preparation of compound **11** was prompted by a report from Glaxo^{2d} that a bromine atom at the 3-position of 2,5-substituted indole or benzofuran AII receptor antagonists was crucial for high potency. Incorporation of bromine into the 1,4-substituted indole nucleus, however, resulted in a loss of potency (**11** vs. **8**).

The compounds in **Table 1** were prepared using chemistry analogous to that outlined in **Scheme 1** which depicts the synthesis of compound **8** (**BMS-180560**). The synthesis began with SNAr coupling of 4-methylindole **12** with 2-fluorobenzonitrile **13** to provide the 1,4-substituted indole **14**. Bromination of indole **14** gave tribromide **15** which was coupled with imidazole **16**⁸ to furnish **17**. Attempted monobromination of **14** led to exclusive introduction of bromine at the indole 3-position. Oxidation of **17** provided the corresponding carboxylic acid which was hydrogenated as a carboxylate salt to yield **18**. Carboxylate formation prior to hydrogenation is a prerequisite since hydrogenation of the protonated acid results in hydrogenolysis of the chlorine on the imidazole. Tetrazole formation on **18** followed by generation of the bis-salt then gave **8** (**BMS-180560**).

Scheme 1. Synthesis of Compound **8** (**BMS-180560**).

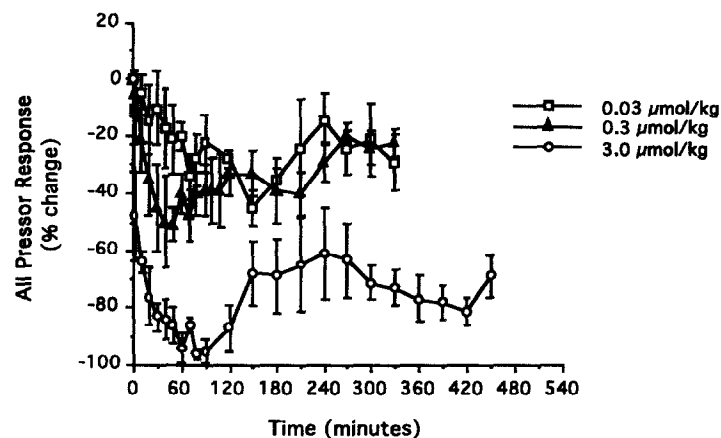


Reagents: a. K_2CO_3 /DMF (85%); b. NBS/ CCl_4 /Benzene (72%); c. *t*-BuOK/*t*-BuOH/DMF (70%); d. $NaClO_2$ / NH_2SO_3H /THF/ H_2O ; e. H_2 /Pearlman's catalyst/ $NaOH/H_2O/EtOH$ (85% for 2 steps); f. *n*- Bu_3SnN_3 /Xylene; g. $LiOH/H_2O/CH_3OH$ (46% for 2 steps).

Compound **8** (**BMS-180560**) was the most potent analog identified in the series. As determined *in vitro* using strips of rabbit aorta,⁵ **8** is a noncompetitive antagonist of the vasoconstrictor effects of AII ($K_B = 0.068 \pm 0.048$ nM). Like many other AII receptor antagonists which contain a carboxylic acid in the imidazole ring,⁹ **8** displays suppression of the maximum effect of AII characteristic of insurmountable antagonism. Radioligand binding studies⁶ demonstrated that **8** selectively binds to the AT_1 population of angiotensin binding sites ($K_i = 0.8 \pm 0.2$ nM for AT_1 , $K_i = 6,500 \pm 3,600$ nM for AT_2).¹⁰

In vivo compound **8** (**BMS-180560**) antagonizes the pressor response to exogenous AII in conscious Sprague-Dawley rats at doses as low as $0.03 \mu\text{mol/kg}$ ($ED_{50} = 0.22 \mu\text{mol/kg}$) when administered i.v. (**Figure 1**). Upon oral administration, at a dose of $30 \mu\text{mol/kg}$ to salt depleted spontaneously hypertensive rats (SHR),¹¹ compound **8** causes a sustained reduction in blood pressure which extends past 72 hours (**Figure 2**). The maximum percent reduction in mean arterial pressure (MAP) is 39% at the $30 \mu\text{mol/kg}$ dose.

Figure 1. Inhibition of the AII Pressor Response in Conscious Sprague-Dawley Rats upon Intravenous Administration of Compound **8** (BMS-180560).



Although compound **8** (BMS-180560) demonstrates reasonable *in vivo* efficacy, it was presumed that its highly polar diacidic character might limit its oral bioavailability. In an effort to improve oral efficacy, prodrug esters of compound **8** were prepared and evaluated in the salt depleted SHR (Table 2). The prodrug esters were prepared by esterification of acid **18** followed by tetrazole formation. Double esters **21-24** demonstrated either marginal or no improvement over acid **8**. The best compound in the series was ethyl ester **19** (BMS-181688). Oral administration of **19** at a dose of 10 $\mu\text{mol/kg}$ to salt depleted SHR causes a sustained reduction in MAP which extends past 72 hours (Figure 2). The maximum percent reduction in MAP is 35% at the 10 $\mu\text{mol/kg}$ dose.

Figure 2. Change in Mean Artrial Pressure in Salt-Depleted SHR upon Oral Administration of Compound **8** (BMS-180560) and Compound **19** (BMS-181688).

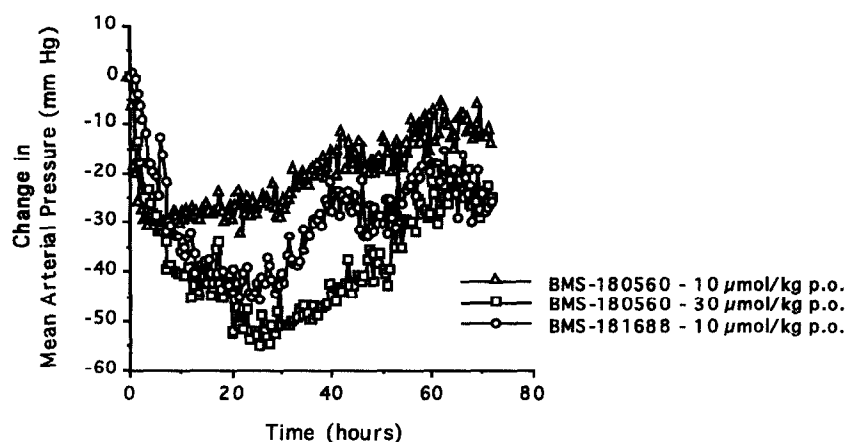
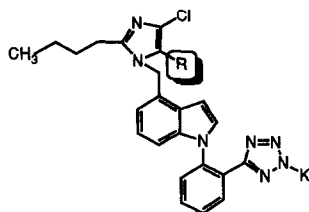


Table 2. Pro-Drug Esters of Compound 8 (BMS-180560).

Compound	R	K _i (nM)	K _B (nM)	Max % b.p. Fall Salt Depleted SHR 10 μmol/kg p.o.
8 (BMS-180560)		0.8	0.068	22% at 18 - 24 hrs
19 (BMS-181688)		25	5	35% at 24 - 30 hrs
20		7.3	6	22% at 18 - 24 hrs
21		4.5	4.5	26% at 6 - 24 hrs
22		30	1.2	29% at 6 - 12 hrs
23		13	4	18% at 24 - 30 hrs
24		8.9	3.6	21% at 24 - 30 hrs
25		6.3	4	15% at 18 - 24 hrs
26		39	130	18% at 24 - 30 hrs

In summary, we have identified a series of 1,4-substituted indoles which are potent, selective, noncompetitive antagonists of the vasoconstrictor effects of Angiotensin II. Compounds in this series are orally active and demonstrate long lasting antihypertensive activity.

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